

230  
geendicee peponde avio, qd ex corbieze rego  
lo no se despi y qd no es menestry.

reci d lo de celo q dren d nra lqz b e conges q  
rest flabada. e se pone en qd la greda fechada qd generada  
q d a g d g d d d d r e e m o y p e g i o q d n a p e d i o  
Or p o d r i n s e g u i z p d b m d e l r o p r o d n e e s t a n d e y  
con g d n e d w o p t u p e d f e c e l v o n i o p d f

1661  
*1661*  
Coyzon al del oba p n o s o n a d e e p e c t i o Q d e y n e  
c r e v i s a g i o y n o s e r e m e d i a . q d n o m d r o l b o m  
z o r i o n m d c o n v e n t o r t i p e s C b m c o n d e s c o h t o  
C l o m o g e s a l v o d g o d a m i s i o n s q n u m e j p a n e  
d e n C u e n t i C H y p r e s G b i z p d e r i b a l v o c e s p o c i e  
C o r a g g c u s o p a s o d e c o d r u m d e

aym /  
m Glunes . xxii d n s d e o p r o v a s d e s p a n a d e f r a n c i a m i t b i n e  
Z o n s t h e o n n e s ) n o c e t r e c / m D o s

| n o c .  
H e a t z  
| m d e s .  
Cas d l D o r e  
Sam / m a /

el m a g . s e n t i m e n t o d i a m p r . ) n e z e f f e s i e n c e ) n o c e m d o r i  
P r o v i d m t r o n f i l m a n g u v a z n e s , k u x c o m t n  
S l a n p /

4 a p o d e l o f r e n t . p l d m u r a & s o n j e z q m d e s m p d e m a d i .  
n o d e a g m i t t e q m d e s m p d e l o f r e n t m a t l n t e l l e s . i d e l o d r  
c o n d o n s o m b e z /  
P l e o s e l o n a q d e l a d u n d e s m i u m . S o b e e l r e g u q d e l a d u n d e  
G l u n d a n g o d s m i s 2 / C r o f n . d l o r d c r p n G r g / m a l t o n t p e  
L e o n n a n e g p o n t r d e a m o n s a t e r i o ( p o r t o ) / S i a f m e a h  
a m o n t u b C e l s p o r a t d o r o a b g d n t r e n e p u d i p a t o d r  
e e a d i m p o p a c f e t n o r e l n e g u a o g v e n g a q d q u i s i e r e p r e c e /  
C q s q n t o a l v o d s r p o d n e t m o d f l o n s e p p n e d e s e r . d n o s o m  
l n g e s u m a g . q y a l o s a b a l a g l h n e / p o o q f i d e m e n e n s -  
i G r e n d a m s e e n / G s e o n g h a a n e p r n q d e m a d i ? @ s e d a c a  
h a n g n t o n a l t z a f a l l o / C o m e t e x o n e l l o m 2 d l o r a l  
s o f r a m d m t r o n y p e

l o g n e l d o s . d l o r a m a /

to home /

Pometeron also  $\text{Fe}^{+2}$  into  $\text{Fe}^{+3}$  and  $\text{Cr}_2\text{O}_7^{2-}$   
Vinegar decomposes  $\text{Fe}_2\text{O}_3$  and  $\text{Cr}_2\text{O}_7^{2-}$  into  $\text{Fe}^{+2}$  and  $\text{Cr}^{+3}$   
Iron hydroxide and aluminum hydroxide are  
soluble in acids.

into

Soluble salt

Reaction with  $\text{H}_2\text{S}$ . Decomposition of  $\text{Fe}^{+2}$  into  $\text{Fe}^{+3}$   
and  $\text{H}_2\text{S}$  forms  $\text{FeS}$ .  $\text{Cr}_2\text{O}_7^{2-}$  reacts with  $\text{H}_2\text{S}$  to form  $\text{Cr}^{+3}$ .  $\text{Cr}^{+3}$  reacts with  $\text{H}_2\text{S}$  to form  $\text{Cr}_2\text{S}_3$ .  $\text{Cr}_2\text{S}_3$  is insoluble in water.  
Iron sulfide and  $\text{Cr}_2\text{S}_3$  are insoluble in water.

Iron sulfide  
is insoluble

Pometeron also  $\text{Fe}^{+2}$  into  $\text{Fe}^{+3}$  and  $\text{Cr}_2\text{O}_7^{2-}$  into  $\text{Cr}^{+3}$   
and  $\text{H}_2\text{S}$  into  $\text{FeS}$  and  $\text{Cr}_2\text{S}_3$ .  $\text{FeS}$  is insoluble in water.  
 $\text{Cr}_2\text{S}_3$  is insoluble in water.  $\text{Cr}_2\text{S}_3$  reacts with  $\text{H}_2\text{S}$  to form  $\text{Cr}_2\text{S}_3 \cdot 2\text{H}_2\text{S}$ .  $\text{Cr}_2\text{S}_3 \cdot 2\text{H}_2\text{S}$  is soluble in water.  
 $\text{Cr}_2\text{S}_3 \cdot 2\text{H}_2\text{S}$  reacts with  $\text{H}_2\text{S}$  to form  $\text{Cr}_2\text{S}_3$  and  $\text{H}_2\text{S}_2$ .  $\text{Cr}_2\text{S}_3$  is insoluble in water.  
Iron sulfide and  $\text{Cr}_2\text{S}_3$  are insoluble in water.

Iron sulfide reacts with  $\text{H}_2\text{S}$  to form  $\text{FeS}$  and  $\text{Cr}_2\text{S}_3$   
in sulfuric acid.  $\text{FeS}$  is insoluble in water.  
 $\text{Cr}_2\text{S}_3$  is soluble in water.

Combustion

Iron decomposes  $\text{H}_2\text{S}$  into  $\text{FeS}$  and  $\text{H}_2$ .  
Iron decomposes  $\text{H}_2\text{S}$  into  $\text{FeS}$  and  $\text{H}_2$ .  
Iron decomposes  $\text{H}_2\text{S}$  into  $\text{FeS}$  and  $\text{H}_2$ .

Combustion of  
iron

Pometeron also  $\text{Fe}^{+2}$  into  $\text{Fe}^{+3}$  and  $\text{Cr}_2\text{O}_7^{2-}$  into  $\text{Cr}^{+3}$   
and  $\text{H}_2\text{S}$  into  $\text{FeS}$  and  $\text{Cr}_2\text{S}_3$ .  $\text{FeS}$  is insoluble in water.  
 $\text{Cr}_2\text{S}_3$  is soluble in water.

Soluble carbon

Iron decomposes  $\text{H}_2\text{S}$  into  $\text{FeS}$  and  $\text{H}_2$ .  
Iron decomposes  $\text{H}_2\text{S}$  into  $\text{FeS}$  and  $\text{H}_2$ .

